

# Experimental and theoretical studies of the ion-pair and valence states of the NeICl van der Waals complexes

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Experimental and theoretical studies of the *T*-shaped NeICl van der Waals (vdW) complexes in the ion-pair (IP)  $E0^+$ ,  $D'2$ ,  $\beta1$  and valence  $A1$  states as well as the NeICl( $A1$ ,  $v_A$ ,  $n_A \leftarrow X0^+$ ,  $v_X = 0$ ,  $n_X$  and  $\beta1$ ,  $v_\beta$ ,  $n_\beta/E_0^+$ ,  $v_E$ ,  $n_E \leftarrow A1$ ,  $v_A$ ,  $n_A$ ) optical transitions have been carried out ( $n_i$  are vdW modes). We have measured NeICl(IP,  $v_{IP} = 0$ ,  $n_{IP} \rightarrow$  valence states) luminescence spectra and their excitation spectra. Binding energies of the states have been determined. The NeICl(IP,  $v_{IP} = 1$ ,  $n_{IP} \rightarrow$  Ne + ICl( $E$ ,  $D'$ ,  $\beta$ )) decay has been also studied. The intermolecular diatomic-in-molecule perturbation theory first order (IDIM PT1) method have been utilized to construct potential energy surfaces (PESs) for the complex valence, and IP states. Calculated spectroscopic characteristics of the NeICl  $E$ ,  $\beta$  and valence states are similar to experimental ones. We calculated energies of the vdW mode and the action NeICl( $A$ ,  $v_A$ ,  $n_A \leftarrow X0^+$ ,  $v_X = 0$ ,  $n_X$ ) as well as excitation NeICl( $E$ ,  $v_E = 0$ ,  $n_E \leftarrow A$ ,  $13$ ,  $n_A$ ), NeICl( $\beta$ ,  $v_\beta = 0$ ,  $n_\beta \leftarrow A$ ,  $13$ ,  $n_A$ ) spectra. Calculated excitation and action spectra describe the principal features of experimental spectra. We achieved satisfactory descriptions of the NeICl( $E$ ,  $0 \rightarrow X$ ), NeICl( $D'$ ,  $0 \rightarrow A'$ ) and NeICl( $\beta$ ,  $0 \rightarrow A$ ) luminescence spectra using Heidelberg MCTDH method, also.

**Keywords:**

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## Introduction

Perturbations of diatomic molecule electronically excited bound states are one of the prime problems of chemical physics and molecular spectroscopy [1]. The complexes, containing rare gas atom and diatomic molecule bonded due to the van der Waals (vdW) interaction, are the simplest system to study and, on the other hand, are convenient model objects for acquiring a knowledge of more complex systems, to study many-body physics.

The study of intermolecular perturbations in optically populated vdW complexes can be facilitated by using dihalogens,  $X_2$ , and interhalogens,  $XY$ , as model systems. Therefore, the rare gas- $X_2$  and rare gas- $XY$  vdW complexes in valence,  $RgX_2(B0_u^+)$ ,  $RgXY(A1, B0^+)$ , states and  $Rg + X_2(B)$  collision-induced non-adiabatic transitions, CINATs, have been studied since 1976 (see Refs. 1–3 and references).

Polarizability which is responsible for the dispersion interaction between Rg atom and dihalogen (interhalogen) in the complex seems to be significantly lower for the ICl molecule compared with the  $I_2$ . However, ICl has a permanent dipole moment (1.207 D for the  $X$  [4]), which is responsible for induction, dipole-induced dipole interaction. Potential energy surfaces (PESs) of the HeICl and ArICl have local minima with similar well depths at *T*-shaped and linear configurations.

Theoretical *ab initio* calculations were performed for the NeICl in the  $X$  [3] and  $B$  [5] state. For the ground

state, it was shown that NeICl( $X$ ) state PES has three minima at linear (NeICl, binding energy  $D_e = 106.8 \text{ cm}^{-1}$ ), near *T*-shaped ( $D_e = 84.79 \text{ cm}^{-1}$ ), and antilinear (ICl-Ne,  $D_e = 76.74 \text{ cm}^{-1}$ ) configurations. O. Roncero *et al.* [5] built PES of the NeICl( $B$ ) state using pairwise potentials. It was shown that there is the only one minimum located in the bent configuration ( $\Theta_e = 140^\circ$ ) with the Ne atoms towards Cl end of ICl.

Experimentally ground  $X0^+$  and excited,  $B0^+$ ,  $A1$  states were studied for the RgICl and RgIBr in the literature, Rg = He, Ne [6–11]. Using the LIF and pump-probe spectroscopy, assignments of the observed bands have been made and the binding energies of the ground and excited states have been estimated (see discussion in Subsection IV.2 of the paper). However, there is discrepancy between estimations of the binding energies.

D.B. Strasfeld *et al* [8] studied NeICl( $B$ ,  $2 - X$ ,  $0$ ) transition. They observed series of bands and assigned them to transitions in different conformers. They attributed a single strong band as a transition in the *T*-shaped complex, and progression of bands as the transition in the linear NeI35Cl one.

RgI $_2$ (IP) vdW complexes have well depth significantly larger as those in the valence electronic states. Giant transition dipole moments between IP states sharing the same  $\Omega$  quantum number lead to the increasing of the  $I_2$  polarizability and increasing of the dispersion interaction (see Section 5 in Ref. 1). The ICl(IP) molecules have not such feature but their permanent dipole moment has to be

larger compared with that in the ground state, and the PES well depths of the  $\text{RgICl(IP)}$  complexes should be large than  $\text{RgICl(valence)}$  ones.

There are several works concerning to  $\text{HeICl}$  [11] and  $\text{NeICl}$  [10,12] complexes in the  $\text{IP} = E0^+, D'2, \beta1$  states. We have studied luminescence of the  $\text{ArICl(IP,vIP)}$  complexes and their decay products in a wide range of the vdW modes. The binding energies of the low vdW levels have been estimated [13–15]. All results show that the PESs of the IP states are deeper than that of the valence one. To determine the  $\text{ArICl}(\beta)$  binding energy, we assumed that the permanent dipole moment of the  $\text{ICl}(\beta)$  molecule is equal to 4 D [15].

Excited valence state  $\text{RgX}_2$ ,  $\text{RgXY}$  complexes undergo fast vibrational and electronic predissociation (VP and EP, respectively). The  $\text{NeICl(A)}$  complex was found to have a long lifetime  $\tau = (3 \pm 2)$  ns for  $\nu_A = 14$  [6] much higher than usual ones for valence complex states (see Sect. 6 in [2] and references). However, lifetimes of the IP low lying vdW states are much longer. The  $\text{NeICl(IP)}$ ,  $\text{IP} = E0^+, D'2$ , and  $\beta1$  complexes were studied by T.A. Stephenson et al. [10]. They found luminescence of the  $\text{ICl(IP)}$  molecules after population of the mutual perturbed  $\text{NeICl}(E, \nu_E = 0-4)$  complex. Because populated vibrational levels lie below any dissociation limits of the IP states, no VP or EP mechanisms are available. They suggested that the origin of the observed luminescence is emission of the  $\text{NeICl(IP)}$  itself, that means at least comparable rates of the radiative and nonradiative decay process.

We determined the lifetime of the  $\text{NeI}_2(E, 0)$  complex with the value  $\tau = 8$  ns due to the complex luminescence and EP [16]. For the  $\text{ArICl}(E, 0)$  complex the measured lifetime  $\tau = 20.3 \pm 1.0$  ns is close to one for free  $\text{ICl(IP)}$  molecules (radiative lifetimes of the free  $\text{ICl}(E, 0, \beta, 0)$ , and  $D', 0$ ) are 22.6(2), 22.6(2) and 14.8(1) (units, ns), respectively) [13]. We explain this value by significant coupling of different „diabatic“  $E, \beta$  and  $D'$  states. It is to be expected the same behavior of the  $\text{NeICl(IP)}$  complexes. However, the lighter Ne atom should not initiate strong coupling of the IP states, and resonance effects should exhibit various coupling of different vdW levels. Therefore, variations of the lifetimes of different vibrational levels is expected to be observed in the  $\text{NeICl}$  complex.

In this paper, theoretical and experimental studies of the  $\text{NeICl(IP, } \nu_{\text{IP}}, n_{\text{IP}})$  at energies lower than the  $\text{NeICl}(E, \nu_E = 0, n_E)$  dissociation limit are presented. In this energy range,  $\text{NeICl(IP, } \nu_{\text{IP}} = 0, n_{\text{IP}})$  VP and EP is unavailable (see Subsection IV.2), and the  $\text{NeICl(IP, } \nu_{\text{IP}}, n_{\text{IP}})$  luminescence can be observed, only [13,14]. Furthermore, excitation and luminescence spectra have also been measured in the energy range higher than the  $\text{NeICl}(E, \nu_E = 0, n_E)$  decay limit where the complex VP and EP are possible.

The first order intermolecular diatomic-in-molecule perturbation theory (IDIM PT1) method [17] is utilized to construct  $\text{NeICl}(A, A'$  and  $E, \beta, D')$  state PESs. Calculated binding energies of the  $T$ -shaped  $\text{NeICl}(A, \text{ and } E, \beta)$  states agree well with experimental ones. Calculated excitation

and action spectra describe the principal features of experimental spectra. We achieved satisfactory descriptions of the  $\text{NeICl}(E, 0, n_X \rightarrow X)$ ,  $\text{NeICl}(D', 0, n_{D'} \rightarrow A')$  and  $\text{NeICl}(\beta, 0, n_\beta \rightarrow A)$  luminescence spectra using Heidelberg MCTDH method, also. These features are supporting the results of calculation.

The complexes containing Ar and Kr atoms are sufficiently difficult for analysis. These heavy atoms perturb all IP states resulting in their strong coupling do not allow to analyze structure of the states in details. On the other hand, well depths of the complexes with light He atom are low and do not have many vibrational levels. Therefore,  $\text{HeICl}$  complex has too simple structure in order to figure out PES features in detail. This work shows advantages of the  $\text{NeICl}$  complex which allows to build detailed PES of the valence and IP states. We have successfully developed a simple but suitable theoretical model.

The paper is organized as following: Methods and procedures utilized in experiments and analysis of experimental data are described in Section II. In Subsection III.1, 2, results of calculations of the complex valence and IP states PESs, their wavefunctions and some spectroscopic characteristics are discussed. An overview of experimental pump-probe and excitation spectra is described in Subsection IV.1. Subsection IV.2 is devoted to direct experimental determination of the complex state binding energies. Comparison of the calculated and experimental pump-probe and excitation spectra is carried out in Subsection IV.3. Subsections IV.4 is devoted to analysis of the luminescence spectra of the  $\text{NeICl(IP)}$  complexes in the energy range lower than their decay limits. Comparison of the experimental  $\text{NeICl(IP)}$  complex luminescence spectra with those calculated using Heidelberg MCTDH method is carried out in Subsection IV.5. Subsections IV.6 is devoted to analysis of the luminescence spectra, population and decay of the  $\text{NeICl(IP)}$  complexes in the energy range higher than their decay limits. In Section V, conclusions are given.

## 1. Experiment and analysis of experimental data

To study the  $\text{NeICl}$  valence and IP states spectroscopic characteristics we utilized a set of methods described in Refs. [13–16,18–26].

To prepare and stabilize  $\text{NeICl(X)}$  we utilized pulsed supersonic jet expansion. The  $\text{He}(99\%) + \text{Ne}(1\%)$  mixture ( $p_{\text{He}} = 10-20$  atm) passed over the bubbler filled with mixed  $\text{ICl}$  and a teflon fillings,  $t \approx 10^\circ\text{C}$ . Two counter-propagating temporally overlapped unfocused TDL90 laser beams crossed the molecular beam at the  $x = (10 \pm 1)$  mm distance from the nozzle with axisymmetric nozzle diameter  $D = 0.8$  mm, so  $x/D \approx 12.5$ . The two-step two-color  $\text{NeICl}$   $\text{NeICl(IP, } \nu_{\text{IP}} = 0, n_{\text{IP}} \xleftarrow{h_2} A, n_A \xleftarrow{h\nu_1} X, 0, n_X)$  scheme was realized.

To distinguish quantum numbers of the populated  $\text{NeCl}(E, v_E/\beta, v_\beta)$  and  $\text{NeCl}(A, v_A)$  vibrational levels and  $n_E, n_\beta, n_A$  vdW modes, luminescence excitation spectra at selected  $\lambda_1, \lambda_{\text{lum}}$  as well as action spectra at selected  $\lambda_2, \lambda_{\text{lum}}$  were measured. We also studied pump-probe spectra ( $v_2$  is fixed at the  $\text{ICl}(E, v_E/\beta, v_\beta \xleftarrow{h\nu_1} A, v_A)$  transitions and  $v_1$  is scanned). Luminescence spectra of the  $\text{NeCl}(\text{IP}, v, v_{\text{IP}}, n_{\text{IP}})$  complexes as well as  $\text{NeCl}(\text{IP}, v_{\text{IP}}, n_{\text{IP}})$  VP and EP products have also been measured. The measurements procedures are reported in Refs. [1,13–16,18–29]. Temporal behaviors of the  $\text{NeCl}(\text{IP}, v_{\text{IP}}, n_{\text{IP}})$  and  $\text{ICl}(\text{IP}, v_{\text{IP}})$  luminescence intensities were also measured.

We used the  $\text{I}^{35}\text{Cl}$ ,  $\text{I}^{37}\text{Cl}$  spectroscopic constants, and RKR curves from Refs. [30,31] ( $A, X$  states), [32–35] ( $E, D', \beta$ ).  $\text{NeI}^{35}\text{Cl}$  isotopomer was studied.

## 2. Theoretical

In this paper, we used the approaches described in our previous work for the  $\text{ArCl}$  and  $\text{HeCl}$  vdW complexes [15,26].

### 2.1. The $\text{NeCl}(X0^+)$ state PES, wave functions and spectroscopic characteristics

The  $\text{NeCl}(X0^+)$  state PES reproduced based on results of *ab initio* calculation carried out in Ref. [3] is shown in Fig. SD1. Fig. 1, *a* shows a comparison of the minimal energy path (MEP) of the  $\text{NeCl}(X)$  PES calculated in this work and in Ref. [3].

They coincide practically. The vdW binding energies and wavefunctions of the  $\text{NeCl}(X)$  state vdW levels are determined using the WavePacket software package in the Matlab environment [36]. In this problem, using the aforementioned calculation package, the stationary Schrödinger equation is solved. Using the discrete variable representation (DVR) for  $R$  and  $\theta$  coordinates [37], the equation is transformed into an eigenvalue problem. This equation is solved by direct diagonalization of the Hamiltonian matrix, resulting in calculated eigenvalues of the Hamiltonian and wave functions in a given coordinate space, the squares of which are usually interpreted as probability densities (see details in Ref. [15,26]). Positions of the  $n_X = 0–3$  vdW levels obtained in this work are similar to those in Ref. 3 (see Table 1) since they have been calculated using  $\text{NeCl}(X)$  state PES of *R. Prosimiti et al* [3].

The calculated values of the  $\text{NeCl}(X)$  state spectroscopic parameters are presented in Table 1. The  $X$  state experimental and calculated binding energies differ significantly.

Plots of the  $J = 0$   $\text{NeI}^{35}\text{Cl}$  probability amplitudes for the  $n_X = 0–2$  vdW modes of the  $\text{NeI}^{35}\text{Cl}(X, v_X = 0)$  state and the same  $\text{NeI}^{35}\text{Cl}(A, 0, \beta, 0$  and  $E, 0)$  vdW modes are given in Fig. 2.

### 2.2. The $\text{NeCl}(A1, \beta1$ and $E0^+)$ state PESs, wavefunctions, and spectroscopic characteristics

The IDIM PT1 method [17] was utilized to construct the  $\text{NeCl}(A, \beta$  and  $E)$  state PESs. The detailed PES building procedure and obtained formulae are described in our previous works [15,26]. The  $\text{NeCl}(A, \beta$  and  $E)$  state PESs and MEPs obtained are given in Fig. SD2 and Fig. 1, *b, c, d*, respectively. The van der Waals binding energies (Table 1) and wavefunction probability amplitudes of the  $\text{NeCl}(A, \beta$  and  $E)$  state vdW levels (Fig. 2) are determined by the way described in previous Section.

Importantly, unlike the  $\text{HeCl}$  system [26], which necessitated further modifications to the PESs, the  $\text{NeCl}$  system only required the inclusion of the induction interaction within the  $\text{Ne-Cl}^-(^1\Sigma^+)$  pair potential [42] using a correction from Ref. [43].

$$V^*(R) = V(R) - f^*(R) \left( \frac{1}{2} \frac{\alpha_1}{R^6} + \frac{1}{2} \frac{\alpha_2}{R^8} \right),$$

where  $\alpha_1$  and  $\alpha_2$  are the dipole and quadrupole polarizabilities of the Ne atom, respectively, and  $f^*(R)$  is a switching function.

$$f^*(R) = \frac{1}{2} [1 + \tanh(1 + \delta R)].$$

The values of  $\alpha_1$  and  $\alpha_2$  for  $\text{Ne-I}(\Pi^3, ^3\Sigma^-)$  were calculated from the pair potentials from Ref. 44, following the methodology outlined in Ref. 15.

$$\alpha_1(\text{Ne} - \text{I}(\Pi^3, ^3\Sigma^-)) = -0.44 \cdot 10^5,$$

$$\alpha_2(\text{Ne} - \text{I}(\Pi^3)) = 0.15 \cdot 10^6,$$

$$\alpha_2(\text{Ne-I}(\Sigma^3)) = -0.13 \cdot 10^6.$$

For  $\text{Ne-Cl}^-(^1\Sigma^+)$ , the values of  $\alpha_1$  and  $\alpha_2$  were determined by optimizing the agreement with experimental data:

$$\alpha_1(\text{Ne-Cl}^-(^1\Sigma^+)) = -0.35 \cdot 10^5,$$

$$\alpha_2(\text{Ne-Cl}^-(^1\Sigma^+)) = 0.114 \cdot 10^6.$$

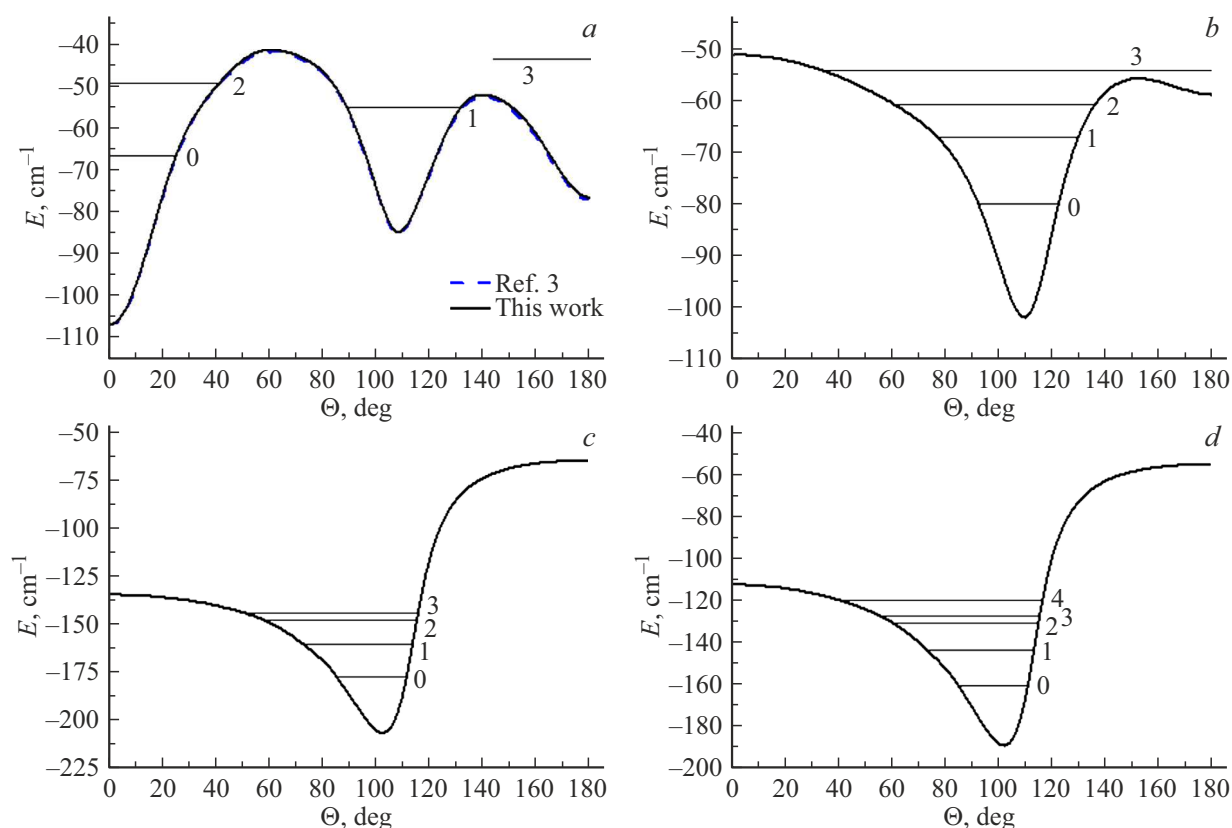
The calculated values, obtained without significant modifications to the PES construction, exhibit good agreement with experimental data.

## 3. Results and discussion

### 3.1. Overview of the pump-probe and excitation spectra

The  $\text{ICl}(E, 0 \rightarrow X$  and  $\beta, 0 \rightarrow A/D', 0 \rightarrow A')$  luminescence pump-probe spectra for pumping in the vicinity of the  $\text{ICl}(A, v_A = 13 \xleftarrow{h\nu_1} X, 0)$  transition and probing at the  $\text{ICl}(\beta, v_\beta = 0 \rightarrow A, 12)$  transition,  $v_2 = 23190.0 \text{ cm}^{-1}$  is given in Fig. 3.

One sees bands corresponding to  $\text{ICl}(A, 13 \leftarrow X, 0)$ ,  $T$ -shaped  $\text{NeCl}(A, 13, n_A \leftarrow X, 0, n_X)$ ,  $T$ -shaped  $\text{Ne}_2\text{ICl}(A,$



**Figure 1.** Minimum energy paths on the  $\text{NeCl}(X)$  (a),  $\text{NeCl}(A)$  (b),  $\text{NeCl}(\beta)$  (c), and  $\text{NeCl}(E)$  (d) obtained in this work and given in Ref. 3. Positions of low vdW levels are shown.

$13 \leftarrow X, 0$ ), and  $T$ -shaped  $\text{Ne}_3\text{ICl}(A, 13 \leftarrow X, 0)$  transitions. Their assignment is made in accordance with the band shift rule (see [45] and references). The  $\text{ICl}(A, 13 \leftarrow X, 0)$  band arises from  $\text{ICl}(A, 13) \xrightarrow{\text{He}} \text{ICl}(A, 12)$  collisions which are still occurring in the observation region. We have observed similar processes many times. The  $\text{NeCl}(A, 13, n_A \leftarrow X, 0, n_X)$  band occurs due to the  $\text{NeCl}(A, 13) \rightarrow \text{Ne} + \text{ICl}(A, 12)$  VP accompanied by transitions to the mixed states  $\text{ICl}(\beta, v_\beta = 0, v_E = 0 \xleftarrow{h\nu_2} A, 12)$  and their luminescence. The  $\text{Ne}_2\text{ICl}(A, 13 \leftarrow X, 0)$  band can be observed (a) due to the  $\text{Ne}_2\text{ICl}(\beta, v_\beta = 2, v_E = 2, n_\beta, n_E \xleftarrow{h\nu_2} A, 13)$  accompanied by VP, EP and luminescence, (b)  $\text{Ne}_2\text{ICl}(A, 13) \xrightarrow{h\nu_2} \text{Ne} + \text{NeCl}(A, 12)$  VP accompanied by the transitions to the mixed states,  $\text{NeCl}(E, D', \beta, v = 0 \xleftarrow{h\nu_2} A, 12)$ , and their luminescence. Similar processes can take place in the  $\text{Ne}_3\text{ICl}$  cluster.

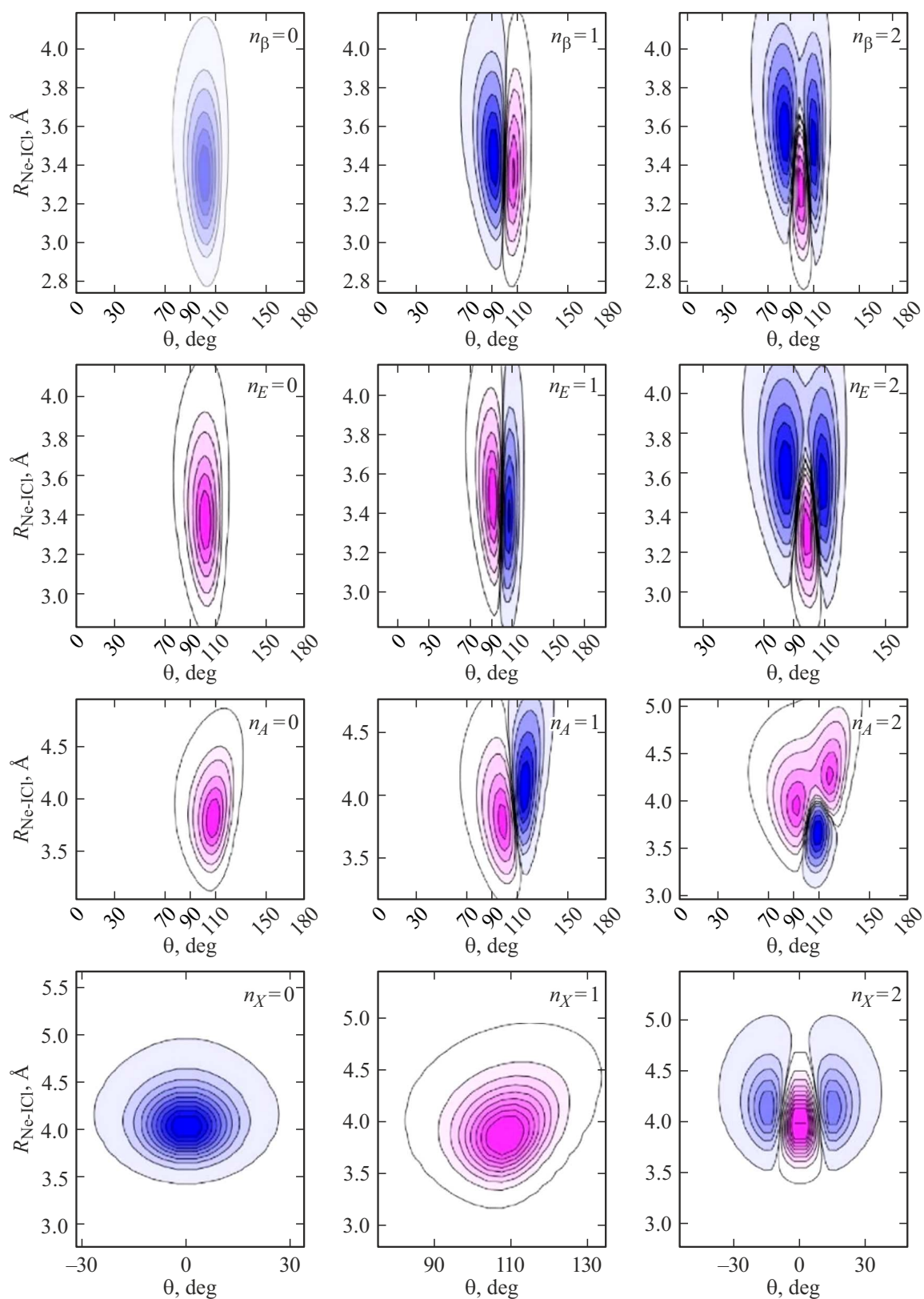
We have observed the  $\text{NeCl}(A, v_A = 13, n_A = 0 \xleftarrow{h\nu_1} X, 0, n_X = 1)$  transition only, in the experiments. All our attempts to detect the linear  $\text{NeCl}(A)$  vdW complexes were fruitless (see Fig. SD3). As it turned out, the wide  $\nu_1^{\text{max}} \approx 15965 \text{ cm}^{-1}$  band belongs to the linear  $\text{HeCl}(A)$  vdW complexes.

The excitation spectra of the  $\text{NeCl}(E, v_E = 0, n_E \rightarrow X)$  and  $\text{NeCl}(\beta, v_\beta = 0, n_\beta \rightarrow A/D', v_{D'} \rightarrow A')$  luminescence for the  $T$ -shaped configuration are given in Fig. 4.

One sees that all the bands in Fig. 4 lie in the energy range lower than the  $\text{NeCl}(\beta, 0, n_\beta)$  dissociation limit,  $23129 \text{ cm}^{-1}$ , and all the bands with the exception of those at  $\nu_2 > 23082 \text{ cm}^{-1}$ , lie in the energy range where  $\text{NeCl}(E, 0, n_E, D', 0, n_{D'})$  decay is impossible. Therefore, luminescence of the  $\text{NeCl}(E, 0, n_E, D', 0, n_{D'}, \beta, 0, n_\beta)$  complexes themselves is only possible in the  $\nu_2 < 23082 \text{ cm}^{-1}$  energy range.

The intensities of excitation bands in Fig. 4 corresponding to the  $\text{NeCl}(\beta, 0, n_\beta \rightarrow A/D', 0, n_{D'} \rightarrow A')$  luminescence are low in the  $\nu_2 < 22950 \text{ cm}^{-1}$  energy range. The  $\text{NeCl}(E, 0, n_E \rightarrow X)$  luminescence is the strongest here, and one may assign the  $\nu_2 = 22922.8$  and  $22941.5 \text{ cm}^{-1}$  bands as belonging to the  $\text{NeCl}(E, 0, n_E = 0, 1 \leftarrow A, 13, n_A = 0)$  transitions. Results of calculations and the  $\text{NeCl}(E, 0, n_E = 0)$  luminescence spectrum confirm this statement (see Sects. IV.3–IV.5).

The first excitation bands at which the  $\text{NeCl}(\beta, 0, n_\beta \rightarrow A/D', 0, n_{D'} \rightarrow A')$  luminescence becomes strong occur at the  $\nu_2 > 22950 \text{ cm}^{-1}$  energy range, and the  $\nu_2 = 22951.8$  and  $22966.6 \text{ cm}^{-1}$  bands may be assigned to the  $\text{NeCl}(\beta, 0, n_\beta = 0, 1 \leftarrow A, 13, n_A = 0)$  transitions. Results of calculations and the  $\text{NeCl}(\beta, 0, n_\beta \rightarrow A/D', 0,$



**Figure 2.** Plots of the  $J=0$   $\text{HeI}^{35}\text{Cl}$  probability amplitudes for the  $n_i=0-2$  vdW modes of the  $\text{NeI}^{35}\text{Cl}(X, v_X=0)$ ,  $\text{NeI}^{35}\text{Cl}(A, v_A=13)$ ,  $\text{NeI}^{35}\text{Cl}(\beta, v_\beta=0)$ , and  $\text{NeI}^{35}\text{Cl}(E, v_E=0)$  states.

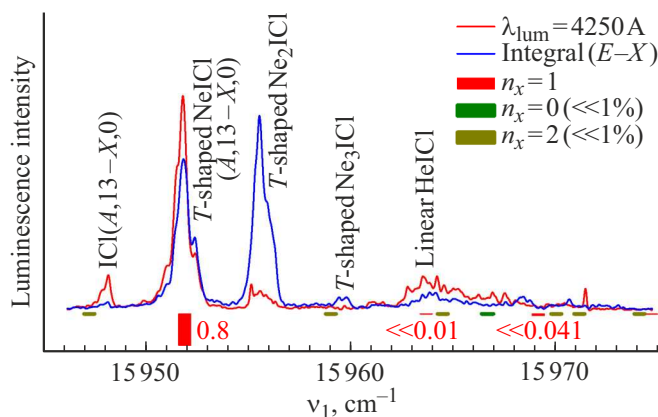
Some spectroscopic parameters (experimental and calculated), potential well depths  $D_e$  ( $\text{cm}^{-1}$ ) and respective equilibrium coordinates  $R_e$  ( $\text{\AA}$ ) and  $\Theta_e$ , binding energies  $D_0$  ( $\text{cm}^{-1}$ ) of the NeICl complexes in the near-T-shaped ( $\Theta \approx 90^\circ$ ) and linear ( $\Theta \approx 0^\circ$ ) configurations (see Sect. 4.2)

State	Reference	Linear, $\theta = 0$		Near T-shaped			
		$D_e$	$D_0$	$R_e$	$D_e$	$D_0$	$R_e/\theta_e$
$X0^+, 0$	[3]	106.8	76.19	3.96	84.79	62.59	3.78/109.1
	[8]		84(1)				
	[38]					70(5)	
	[39]					70	
	[12]					48.2(5)	
A1	This work, calculated	106.8	79.62	3.96	84.7	64.86	3.77/109.1
	Experiment		$\sim 100$			84(1)	
	[10]					$\leq 60^{(1)}$	
	[12]		48.2(5)			43–45	
	[40,41]					41–45	
$\beta 1$	[38]					62–65 <sup>(2)</sup>	3.7/109.4
	This work, calculated				101.8	80.37 <sup>(3)</sup>	
	Experiment					80(1) <sup>(3)</sup>	
	This work, calculated				205.4	178.06	3.3/102.5
	Experiment					178(1)	
$E0^+$	[12]					87.6(8)	
	This work, calculated				189.1	161.03	3.3/102.5
	Experiment					161(1)	

Note. <sup>(1)</sup>  $v_A = 14$ ,

<sup>(2)</sup>  $v_A = 23$ ,

<sup>(3)</sup>  $v_A = 13$ .



**Figure 3.** The pump-probe spectra of the  $\text{ICl}(E, 0 \rightarrow X$  and  $\beta, 0 \rightarrow A/D', 0 \rightarrow A')$  luminescence in the vicinity of the  $\text{ICl}(A, v_A = 13 \xleftarrow{h\nu_1} X, 0)$  transition and probing at the  $\text{ICl}(\beta, v_\beta = 0 \xleftarrow{h\nu_2} A, 12)$  transition,  $v_2 = 23\,190.0\text{ cm}^{-1}$ . Franck-Condon factors of the  $\text{NeICl}(A, v_A = 13, n_A \xleftarrow{h\nu_1} X, 0, n_X = 0, 1, 2)$  transitions calculated using wave functions obtained in this work (see Fig. 2) are presented.

$n_{D'} \rightarrow A'$  luminescence spectra confirm this statement (see Sects. IV.3-IV.4).

Comparison of excitation spectra corresponding to the population of  $v_{E,\beta} = 0, 1$  vibronic levels of the complexes is shown in Fig. 5.

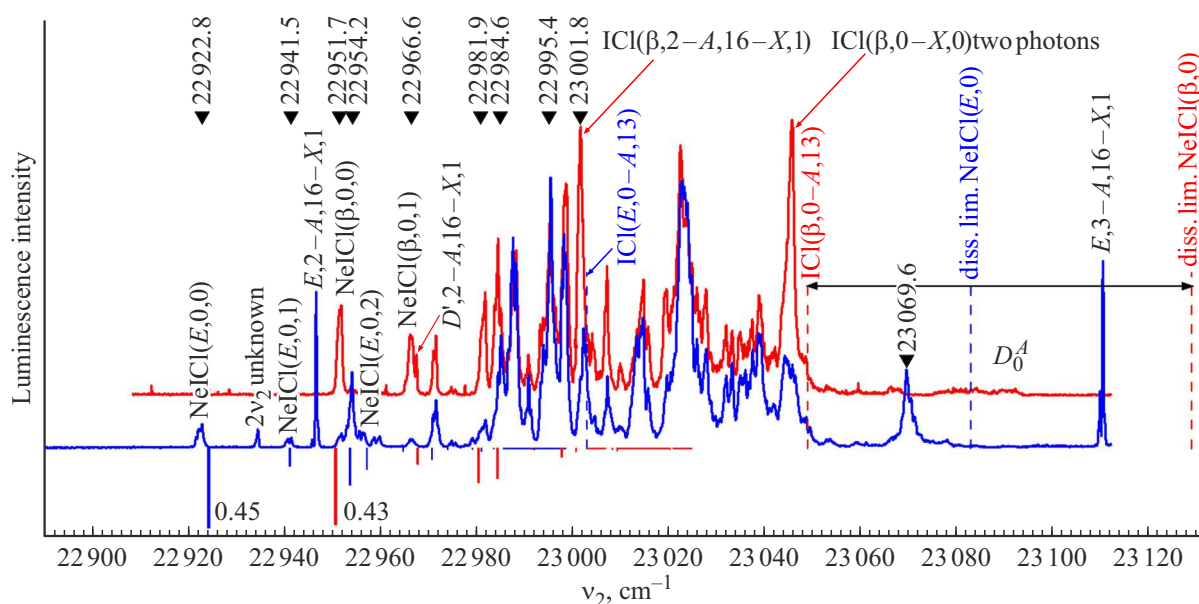
The  $\text{ICl}(\beta, 1 - A, v_A = 13)$  FCF is low, 0.0036. Therefore, the  $\text{NeICl}(A, v_A = 12, n_A = 0)$  level has been used as intermediate one for population of the  $\text{HeICl}(\beta, 1, n_\beta)$  complexes (unfortunately, FCF of the  $\text{ICl}(E, 1 - A, v_A = 12)$  transition is also low, 0.004). One should note that  $\text{NeICl}(E, \beta, 1, n_{E,\beta})$  complexes can undergo EP and VP.

Positions of the excitation band in Fig. 5 are similar but do not coincide. Differences lie in the  $\Delta(v_1 + v_2) = 0 - 3.5\text{ cm}^{-1}$ ; the  $\Delta(v_1 + v_2) = 3.5\text{ cm}^{-1}$  is the energy difference of the  $\text{NeICl}(\beta, v_\beta = 0, 1, n_\beta = 0 \leftarrow A, 13, n_A = 0)$  transitions. The  $\text{ICl}(\beta, 1 - \beta, 0)$  energy difference is  $169\text{ cm}^{-1}$ , and  $\text{NeICl}(\beta, v_\beta = 1, n_\beta = 0)$  binding energy should be  $178\text{ cm}^{-1}$  (see Table 1). Therefore, the  $\text{NeICl}(\beta, v_\beta = 1, n_\beta = 0)$  level, for example, lies a little below the  $\text{NeICl}(\beta, v_\beta = 0, n_\beta)$  dissociation limit and may be perturbed by a high  $\text{NeICl}(\beta, v_\beta = 0, n_\beta)$  levels.

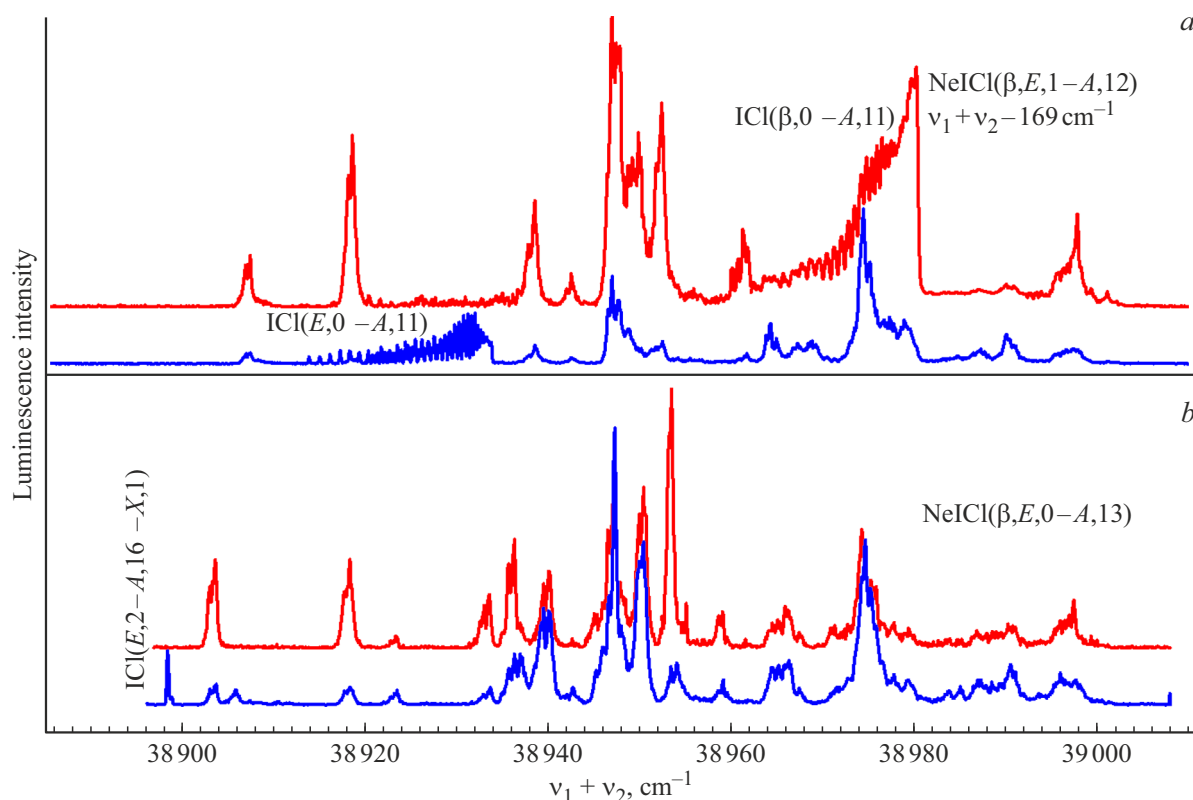
Excitation spectra measured in Ref. [10] and this work are given in Fig. SD4. Positions of the  $\text{NeICl}(\beta \rightarrow A/D' \rightarrow A')$  bands in Fig. SD4, *a, b* are similar. As to the  $\text{NeICl}(E \rightarrow X)$  bands, Stephenson *et al* did not observe them in the  $v_1 + v_2 < 38\,940\text{ cm}^{-1}$  spectral range due to the low signal-to-noise ratio, probably.

### 3.2. Determination of the NeICl state binding energies

Data on  $\text{NeICl}(X, A, E, \beta)$  state binding energies and their analysis are presented in Refs. [3,8,10,12,38,41]. In



**Figure 4.** The  $\text{NeCl}(E, \beta, \nu_E, \beta = 0, n_{E,\beta} \leftarrow A, 13, n_A = 0)$  excitation spectra of the luminescence measured in the  $\lambda_{\text{lum}} = 3500\text{--}4000\text{ \AA}$  spectral range (blue line) and at  $\lambda_{\text{lum}} = 4250\text{ \AA}$  (red line),  $\nu_1 = 15951.8\text{ cm}^{-1}$ ,  $T$ -shaped  $\text{NeCl}(A, 13, n_A = 0 \leftarrow X, 0, n_X = 1)$  transition. Positions of the bands at which luminescence spectra have been measured are marked by triangles. Dissociation limits of the  $\text{NeCl}(\beta, 0)$  and  $\text{NeCl}(E, 0)$  complexes are marked, and calculated FCFs of the  $\text{NeCl}(E, \nu_E = 0, n_E \leftarrow A, 13, n_A)$ ,  $\text{NeCl}(\beta, \nu_\beta = 0, n_\beta \leftarrow A, 13, n_A)$  transitions are shown.



**Figure 5.** Excitation spectra of the luminescence measured in the  $\lambda_{\text{lum}} = 3500\text{--}4000\text{ \AA}$  spectral range (blue line) and at  $\lambda_{\text{lum}} = 4250\text{ \AA}$  (red line) using the  $\text{NeCl}(E, \beta, 1, n_{E,\beta} \xleftarrow{h\nu_2} A, \nu_A = 12, n_A = 0 \xleftarrow{h\nu_1} X, 0, n_X = 1)$ ,  $\nu_1 = 15809.5\text{ cm}^{-1}$  (a), and  $\text{NeCl}(E, \beta, 0, n_{E,\beta} \xleftarrow{h\nu_2} A, \nu_A = 13, n_A = 0 \xleftarrow{h\nu_1} X, 0, n_X = 1)$ ,  $\nu_1 = 15951.8\text{ cm}^{-1}$  (b) schemes. The  $\nu_2$  generation intensity is  $\sim 15$  times lower than that of using in Fig. 4. The  $\text{ICl}(\beta, 1 - \beta, 0)$  energy difference,  $169\text{ cm}^{-1}$ , is subtracted from the  $\nu_1 + \nu_2$  value in the a panel.



all experimental works, the highest rotational levels of the  $\text{ICl}(A, v_A, J_A)$  states nascent in  $\text{NeICl}(A)$  VP,  $\text{ICl}(A, v_A, J_A^{\max})$ , were used for the determination of the  $\text{ICl}(A, v_A, J_A)$  state rotational excitation energy. Then the  $\text{NeICl}(X, A, E, \beta)$  binding energies were determined using the spectral shifts of the  $\text{NeICl}(E/\beta \leftarrow A)$  and  $\text{NeICl}(A \leftarrow X)$  transitions relative to those of a free molecule.

However, vibrational energy of the complex in VP releases also into kinetic energy of fragments recoil, and the energy of the  $\text{NeICl}(A, v_A, n_A)$  is equal to the sum of  $\text{ICl}(A, v_A - 1, J_A^{\max})$  energy and the kinetic energy of the  $\text{Ne} + \text{ICl}(A, v_A, J_A)$  fragments (kinetic energy release, KER).<sup>13</sup> In the case of the  $\text{ArICl}(A, v_A, n_A)$  predissociation,  $187 \text{ cm}^{-1} \leq D_0^{A, v_A=13} \leq 210 \text{ cm}^{-1}$  [13]. One sees that determination error of the  $\text{NeICl}(A, v_A, n_A)$  binding energy is large.

Therefore, we decided to carry out its direct measurement in the manner used in Ref. [8,45], i.e. to measure pump-probe,  $\text{NeICl}(A, 13, n_A \leftarrow X, 0, n_X)$  (pump),  $\text{ICl}(E, 0 \leftarrow A, 13)$  (probe) spectra, and find the  $\text{NeICl}(A, 13, n_A)$  dissociation limit. We knew that  $\text{NeICl}(A, 13, n_A \leftarrow X, 0, n_X)$ ,  $\text{ICl}(E, 0 \leftarrow A, 13)$  transitions had to exist, and  $D_0^{A, v_A} \geq 60 \text{ cm}^{-1}$ . Therefore, we used the  $\text{Ne}(5\%) + \text{He}(95\%)$  mixture and carried out measurements in large,  $\nu_1 \approx 15940\text{--}16050 \text{ cm}^{-1}$  energy range. One sees in Fig. SD3 the bands corresponding to populations of the  $\text{ICl}(A, 13)$  free molecule,  $\text{NeICl}(A, 13, n_A = 0)$   $T$ -shaped complex,  $\text{Ne}_n\text{ICl}(A, 13, n_A = 0)$ ,  $n = 2, 3$  clusters, and  $\text{HeICl}(A, 13, n_A = 1)$  linear (free-rotor) complex. According to results of our calculations, the intensities of  $\text{NeICl}(A, 13, n_A = 1, 2 \leftarrow X, 0, n_X)$  transitions are negligible (see Sect. IV.2).

One should discuss which, the  $T$ -shaped or linear (free-rotor)  $\text{NeICl}(X)$  complexes, corresponds to this threshold. Both linear  $\text{NeICl}(X, 0, n_X = 0)$  and  $T$ -shaped  $\text{NeICl}(X, 0, n_X = 1)$  complexes exist in a supersonic beam. One cannot observe  $\text{ICl}(A)$  luminescence and determine its rotational temperature due to large  $\text{ICl}(A)$  radiative lifetime,  $> 10^{-4} \text{ s}$  (see Ref. [46]). Therefore, we decided to use  $\text{ICl}(X, 0)$  rotational temperature,  $T_{\text{rot}} = 1.36(2) \text{ K}$ , determined in Ref. [47] for  $\text{HeICl}$  in the conditions ( $p_{\text{He}} = 18.5 \text{ bar}$ ,  $x/D = 11.9$ ) very similar to those used in our experiments ( $p_{\text{He}} = 20 \text{ atm}$ ,  $x/D = 12.5$ ). Population of the  $\text{NeICl}(X, 0)$   $n_X = 1$  ( $T$ -shaped) mode is  $\exp(-15/5.4)$  less than that of  $n_X = 0$  (linear) one ( $15 \text{ cm}^{-1}$  and  $5.4 \text{ cm}^{-1}$  are  $\text{NeICl}(X, 0)$  and  $\text{HeICl}(X, 0)$  energy differences of the  $n_X = 0$  and  $1$  modes, respectively) [11]. Therefore,  $\text{NeICl}(X, 0, n_X = 1)$  population is  $\exp(-15/5.4) \cdot (0.37/0.65) = 0.036$  of that of  $\text{NeICl}(X, 0, n_X = 0)$  ( $0.65$  and  $0.37$  are relative populations of the  $\text{NeICl}(X, 0)$   $n_X = 0$  and  $1$  conformers at  $x/D = 13$ ) [48]. Nevertheless, we have not observed the linear  $\text{NeICl}(A)$  conformer due to negligible FCFs of the linear  $\text{NeICl}(X, 0, n_X = 0)$  conformer transitions.

According to the results of our calculations, there is no minimum at the  $\text{NeICl}(A, 13, n_A)$  PES at linear configuration (see Fig. SD2), the  $n_A \leq 2$  vdW levels corresponds to the near  $T$ -shaped configuration (Fig. 1,  $b$ ). The  $\text{NeICl}(A,$

$13, n_A \geq 5 \leftarrow X, 0, n_X = 0$ ) transitions are of negligible probabilities due to large differences of  $R$  values corresponding to the  $\text{NeICl}(X, 0, n_X = 0)$  ( $3.97 \text{ \AA}$ ) and  $\text{NeICl}(A, 13, n_A \geq 5)$  ( $\sim 4.5 \text{ \AA}$ ) states. Franck-Condon transition from the linear  $\text{NeICl}(X, 0, n_X = 0)$  ( $R = 3.97 \text{ \AA}$ ) state occurs at the repulsive part of the  $\text{NeICl}(A, 13, n_A)$  PES lying much higher than the  $\text{NeICl}(A, 13, n_A)$  dissociation limit. Therefore, the  $\nu_1 = 16032 \text{ cm}^{-1}$  threshold corresponds to the  $T$ -shaped configuration, and the  $\text{NeICl}(A, 13, n_A = 0)$  binding energy is equal to  $D_0^{A, 13} = 80(1) \text{ cm}^{-1}$ ,  $\sim 15 \text{ cm}^{-1}$  larger than determined earlier [10,12,38,40,41]. The  $\text{NeICl}(X, 0, n_X = 1)$  binding energy is equal to  $D_0^{X, 0, 1} = 84(1) \text{ cm}^{-1}$  since the  $\text{ICl}(A, 13) \rightarrow \text{NeICl}(A, 13, n_A = 0)$  blue shift is  $4 \text{ cm}^{-1}$  and the linear  $\text{NeICl}(X, 0, n_X = 0)$  binding energy is  $D_0^{X, 0} \approx 100 \text{ cm}^{-1}$  (see Table 1, Fig. 6).

Energy level diagram for the  $\text{NeICl}(X, 0, n_X = 1, A, 13, n_A = 0, E, 0, n_E = 0, \beta, 0, n_\beta = 0)$  complexes calculated using the  $T$ -shaped  $\text{NeICl}$  binding energies,  $D_0^{A, 13} = 80(1) \text{ cm}^{-1}$ ,  $D_0^{X, 0} = 84(1) \text{ cm}^{-1}$ , determined, and other experimental data obtained in this work is given in Fig. 6.

One should note that  $\text{Ne} + \text{ICl}(E, 0)$  dissociation limit lies  $161 \text{ cm}^{-1}$  higher than the  $\text{NeICl}(E, 0, n_E = 0)$  state. Dissociation limits of the  $T$ -shaped  $\text{NeICl}(D', 0)$  and  $\text{NeICl}(\beta, 0)$  complexes are  $1.4 \text{ cm}^{-1}$  and  $46.4 \text{ cm}^{-1}$  higher (see Fig. 4, also). Therefore,  $\text{NeCl}(\text{IP}, \nu_{\text{IP}} = 0, n_{\text{IP}} = 0)$  VP and EP is unavailable.

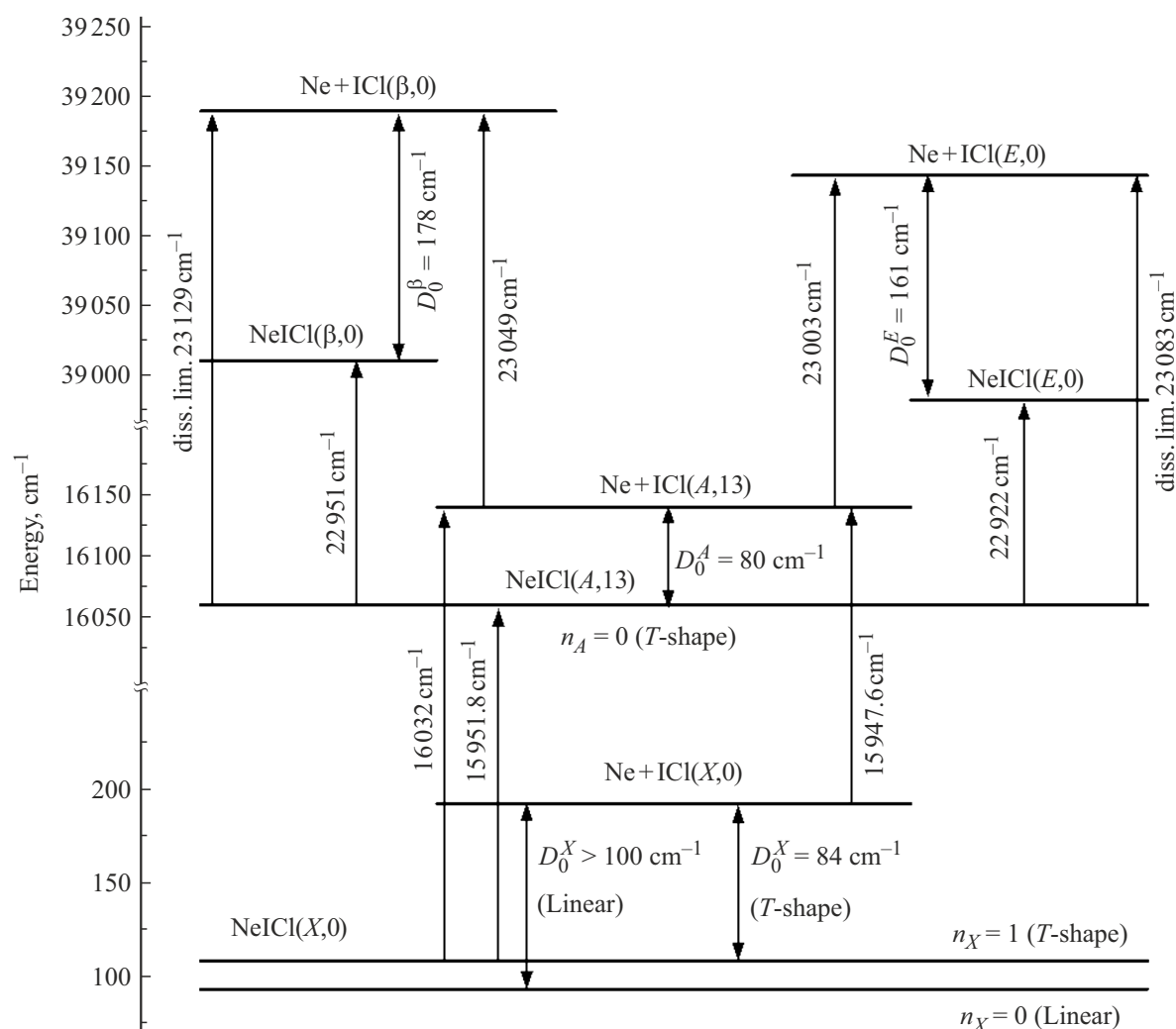
In principle,  $\nu_1 > 16032 \text{ cm}^{-1}$  continuum can correspond to the  $\text{Ne}_2\text{ICl}(E, 0, n_E) \rightarrow \text{Ne} + \text{NeICl}(E, 0, n_E)$  cluster dissociation. It is well known, however, that  $\text{RgICl}(E, 0, n_E \rightarrow X, \nu_X)$  luminescence bands are similar to those of  $\text{ICl}(E, 0, n_E \rightarrow X, \nu_X)$  but are shifted relative to the latter (see Refs. [2,26] and references). The luminescence spectrum measured at the continuum coincides with the  $\text{ICl}(E, 0 \rightarrow X)$  one (Fig. SD5).

The data on the  $\text{NeICl}(X, A, E)$  binding energies obtained in this work differ significantly from those given in Refs. [3,8,10,12,38–41] (see Table 1). We have to discuss these contradictions.

D.B. Strasfeld et al [8] have measured LIF,  $\text{I}^{35}\text{Cl}(B, \nu_B = 2 \rightarrow X, \nu_X)$  and pump,  $\text{NeI}^{35}\text{Cl}(B, \nu_B = 2 \rightarrow X, \nu_X)$ , probe,  $\text{I}^{35}\text{Cl}(E, \nu_E = 11 \leftarrow B, \nu_B = 2)$ , spectra at different  $x/D = 7.5\text{--}25$  ( $x$  is the downstream distance and  $D$  is the nozzle diameter). They have observed the single strong band corresponding to the  $T$ -shaped complex, and progression corresponding, as authors believe, to the linear  $\text{NeI}^{35}\text{Cl}(B, \nu_B = 2)$  complexes in LIF spectra. Integral intensity of the  $T$ -shaped band is larger than total intensity of the linear bands. Besides, they observed continua in LIF and pump-probe spectra corresponding, as authors believe, to transitions of the linear  $\text{NeI}^{35}\text{Cl}(X, \nu_X = 0)$  conformer to above the  $\text{Ne} + \text{I}^{35}\text{Cl}(B, \nu_B = 2)$  asymptote.

The authors of Ref. [8] have affirmed that the continuum fluorescence intensity in the LIF spectra tracks with the linear progression intensity when  $x/D$  increase from  $x/D = 7.5$  to  $25$ . Therefore, observed results from transitions of the linear  $\text{NeI}^{35}\text{Cl}(X, 0)$  conformer.





**Figure 6.** Energy level diagram for the NeICl( $X, 0, n_X, A, 13, n_A = 0, E, 0, n_E = 0, \beta, 0, n_\beta = 0$ ) complexes.

There are serious doubts about the correctness of the assignment the progression and continua to the linear conformer.

1. According to the calculations of O. Roncero et al [5], there is the only one minimum at NeICl( $B, v_B$ ) PES, and it is located in the bent configuration,  $D_e = -67 \text{ cm}^{-1}$ ,  $R_e = 4.95 \text{ \AA}$ ,  $\Theta_e = 140^\circ$  (the Ne atoms towards Cl end of ICl). No minimum exists in the linear configuration,  $\Theta = 0^\circ$ , and bound-bound NeICl( $B, v_B = 2 \leftarrow X, v_X$ ) transitions are impossible in the linear configuration.

2. One sees in Fig. 2 of Ref. [8] that the average ratio of the intensities of the linear conformer bands measured at a rotational temperature of 0.81 K to those of at 3.9 K is 1.8, while for the continuum this ratio is 1.4. If the continuum belongs to the linear conformer, these ratios should be the same.

3. We have measured the pump-probe, NeICl( $A, 13, n_A \leftarrow X, 0, n_X$ ) (pump), ICl( $E, 0 \leftarrow A, 13$ ) (probe,  $\nu_2 = 23003 \text{ cm}^{-1}$ ) spectrum,  $\lambda_{\text{lum}} = 3500\text{--}4000 \text{ \AA}$ , ICl( $E \rightarrow X$ ) transition in wide,  $15938\text{--}16051 \text{ cm}^{-1}$  spectral range (see Fig. SD3). If the continuum observed in

our experiments belongs to the linear conformer, the continuum belonging to the  $T$ -shaped conformer has to start at  $\sim (16032 - 15) = 16017 \text{ cm}^{-1}$ . We have observed no continuum at  $\nu_2 > 16017 \text{ cm}^{-1}$ . There is only a weak continuum started at  $\nu_1 = 15970 \text{ cm}^{-1}$  and corresponding to transitions in the HeICl complex. It should also be noted that, in the HeICl vdW complex. these two continua are observed [45], and the difference in the wave number of their appearance is equal to  $3 \text{ cm}^{-1}$ , i.e., the difference in the binding energies of the linear and  $T$ -shaped conformers of NeICl( $X$ ) [3].

T.A. Stephenson et al [10] using the data of Ref. [41] (NeICl( $A, v_A = 23$ )) and conception of Ref. [38] (see below) estimated the NeICl( $A, v_A = 14$ ) binding energy as  $\leq 60 \text{ cm}^{-1}$ .

K.C. Janda, C.R. Bieler [38] estimated the NeICl( $X, 0$ ) binding energy as  $70(5) \text{ cm}^{-1}$  using the NeICl( $A, v_A = 23$ ) VP product rotational distribution obtained in Ref. [41].

J.I. Cline et al [39] proposed the  $T$ -shaped  $D_o(\text{NeICl}(X)) = 70 \text{ cm}^{-1}$  comparing the NeCl<sub>2</sub> dynamics to those of NeICl.

J.C. Drobits, M.I. Lester [12] measured double resonance spectrum by promoting  $\text{NeICl}(X, 0)$  to  $\text{NeICl}(A, \nu_A = 12, 14, 15)$  and probing  $\text{NeICl}(E, \nu_E = 1 - A, \nu_A)$  transition. They determined the  $\text{NeICl}(E, \nu_E = 1)$  binding energy as 87.6(8) after analysis of progression measured and then those of  $\text{NeICl}(X, 0)$  and  $\text{NeICl}(A, \nu_A = 12, 14, 15)$  using pump and probe blue shifts.

One sees in Table 1 that discrepancies of the binding energies obtained earlier are too large.

The authors of Ref. [40,41] believe that binding energies of the  $\text{NeICl}(A, \nu_A = 11, 14, 19, 23)$  complexes lies in the  $D_0^{A,11-23} \approx 41-45 \text{ cm}^{-1}$  range, whereas authors of Ref. [38] believe that  $D_0^{A,23} = 62-65 \text{ cm}^{-1}$ . As to J.C. Drobits and M.I. Lester paper [12], their assignment of the bands and especially the dissociation limit in Fig. 4 of the paper questionable. The peak assignment by the author to continuum is too narrow to be a continuum (see Fig. SD3). We measured similar spectra (Figs. 4, 5), but we found two additional bands in the  $\text{NeICl}(E, \beta, \nu_{E,\beta} = 0, n_{E,\beta} \leftarrow A, 13, n_A = 0)$  excitation spectra of the luminescence, lying deeper, at  $\nu_2 = 22922.8$  and  $22941.5 \text{ cm}^{-1}$  (Fig. 4). Moreover, no continua were observed in our spectra at  $\nu_2 > (22922.8 + 87.6) = 23010.4 \text{ cm}^{-1}$ . Therefore, the values of  $\text{NeICl}(E, \nu_E = 1)$  binding energies,  $87.6 \text{ cm}^{-1}$ , determined in Ref. [12] are underestimated.

As is mentioned above, we have not observed the linear conformer in the pump-probe spectra of the  $\text{ICl}(E, 0 \rightarrow X$  and  $\beta, 0 \rightarrow A/D', 0 \rightarrow A')$  luminescence in the vicinity of the  $\text{ICl}(A, 13 \xleftarrow{h\nu_1} X, 0)$  transition and probing at the  $\text{ICl}(\beta, \nu_\beta = 0 \xleftarrow{h\nu_2} A, 13)$  transition (see Fig. SD3). According to our calculation,  $\text{NeICl}(A, 13, n_A \geq 5 \leftarrow X, 0, n_X = 0)$  transitions (linear conformer) are of negligible probabilities due to large differences of  $R$  values corresponding to the  $\text{NeICl}(X, 0, n_X = 0)$  ( $3.97 \text{ \AA}$ ) and  $\text{NeICl}(A, 13, n_A \geq 5)$  ( $\sim 4.5 \text{ \AA}$ ) states. Franck-Condon transition from the linear  $\text{NeICl}(X, 0, n_X = 0)$  ( $R = 3.97 \text{ \AA}$ ) state occurs inevitable at the repulsive part of the  $\text{NeICl}(A, 13, n_A)$  PES lying much higher than the  $\text{NeICl}(A, 13, n_A)$  dissociation limit. Therefore, according to our data, the strong continuum in Fig. SD3 in the manuscript new version cannot be assigned to the linear conformer.

### 3.3. Comparison of the calculated and experimental pump-probe and excitation spectra

Comparison of experimental and calculated pump-probe spectra is presented in Fig. 3. The  $n_X = 0, 2$  and 1 vdW modes are located in the linear ( $\text{Ne} \cdots \text{I-Cl}$ ), antilinear ( $\text{Ne} \cdots \text{I-Cl}$ ) and near  $T$ -shaped configurations, respectively (see Figs. 1, 2 and Ref. [3]). The low,  $n_A = 0-2$ , vdW modes are located in near  $T$ -shaped configuration, and cannot be populated from the linear  $n_X = 0$  mode. Calculations confirm this statement, and experimental pump-probe spectra agree very well (Fig. 3).

Comparison of experimental and calculated  $\text{NeICl}(E, 0, n_E, \beta, 0, n_\beta \xleftarrow{h\nu_2} A, 13, n_A = 0)$  excitation spectra is presented in Fig. 4. According to the results of calculations, the  $\text{NeICl}(E, 0, n_E = 0 \leftarrow A, 13, n_A = 0)$  and  $\text{NeICl}(\beta, 0, n_\beta = 0 \leftarrow A, 13, n_A = 0)$  transitions are the strongest, and this feature corresponds to the probability amplitudes given in Fig. 2. It is impossible to compare calculated FCFs and intensities of experimental excitation bands since laser generation intensity increases with  $\nu_2$  value in the  $\nu_2 = 22920-23020 \text{ cm}^{-1}$  spectral range, and we have not measured this intensity. Besides, there are power saturation at some bands (see Figs. 4 and 5, b). Positions of the calculated and experimental bands are very similar.

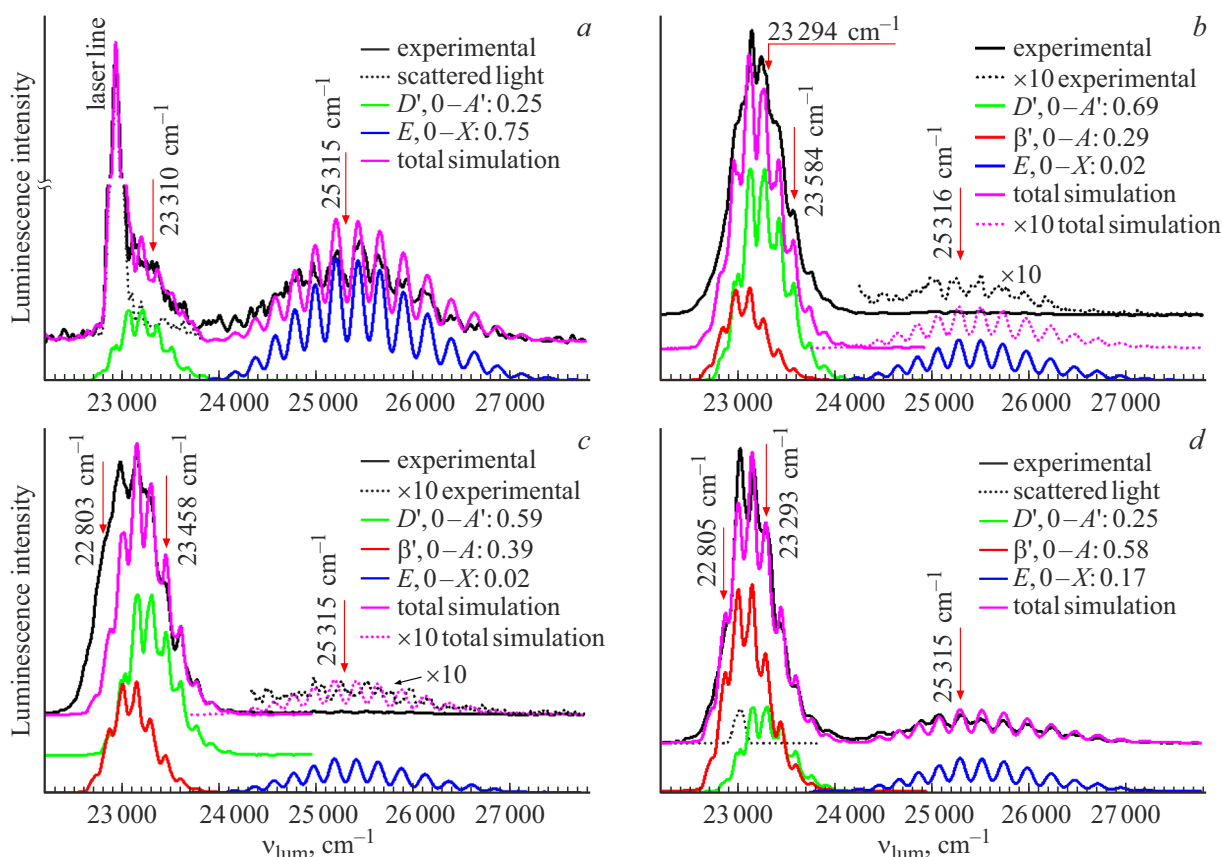
### 3.4. Analysis of the luminescence spectra and population and decay of the $\text{NeICl}(\text{IP})$ complexes in the energy range lower than their decay limits

Luminescence spectrum measured at the least wavenumber observed in excitation spectra ( $22922.8 \text{ cm}^{-1}$  in Fig. 4) and assigned by us to the  $\text{NeICl}(E, 0, 0 \leftarrow A, 13, 0)$  transition is given in Fig. 7, a.

We tried to describe the spectrum given in Fig. 7, a by luminescence spectra of a free  $\text{ICl}$  molecule (similar attempts were undertaken for Figs. 7, b-d). We have described the  $\nu_{\text{lum}}^{\text{max}} \approx 25200 \text{ cm}^{-1}$  band using the  $\text{ICl}(E, 0 \rightarrow X)$  luminescence spectrum shifted by  $140 \text{ cm}^{-1}$  to the red. To describe the luminescence at the  $\nu_{\text{lum}}^{\text{max}} \approx 23000 \text{ cm}^{-1}$  band we have to use the  $\text{ICl}(D', 0 \rightarrow A')$  spectra shifted by  $140 \text{ cm}^{-1}$  to the red. However,  $140 \text{ cm}^{-1}$  red shift is larger than that obtained from Fig. 6 for  $\text{NeICl}(E)$ ,  $80 \text{ cm}^{-1}$ .

Luminescence spectrum measured at  $\nu_2 = 22951.8 \text{ cm}^{-1}$  where the  $\text{NeICl}(D', 0, n_{D'} \rightarrow A'$  and  $\beta, 0, n_\beta \rightarrow A)$  transitions are much stronger than the  $\text{NeICl}(E, 0 \rightarrow X)$  one (see Figs. 5 and 7, b) are described by the shifted  $\text{ICl}(D', 0 \rightarrow A', \beta, 0 \rightarrow A)$  and  $\text{ICl}(E, 0 \rightarrow X)$  spectra satisfactorily. One may not say the same about the spectrum measured at  $\nu_2 = 22966.3 \text{ cm}^{-1}$  (Fig. 7, c) since its low-frequency part cannot be described by the shifted  $\text{ICl}(D', 0 \rightarrow A'$  and  $\beta, 0 \rightarrow A)$  spectra. As to the spectrum measured at  $\nu_2 = 22995.4 \text{ cm}^{-1}$ , it is described by the shifted  $\text{ICl}(D', 0 \rightarrow A', \beta, 0 \rightarrow A)$  and  $\text{ICl}(E, 0 \rightarrow X)$  spectra satisfactorily, again. One sees that the luminescence intensity temporal behaviors depend on excitation wavenumbers (Fig. SD6). Besides, at the  $\nu_2 = 22951.8$  and  $22966.3 \text{ cm}^{-1}$  bands, maxima of the  $\text{NeICl}(E, 0, n_E \rightarrow X)$  temporal behaviors are shifted to larger time relative to those measured at  $\nu_{\text{lum}}^{\text{max}} \approx 23000 \text{ cm}^{-1}$ , as it occurs in the case of slow non-adiabatic transitions (see Ref. [29]).

The  $E \rightarrow X$  luminescence intensity at the  $22951.8 \text{ cm}^{-1}$  and  $22966.8 \text{ cm}^{-1}$  bands (Fig. SD 6, b, c) is approximately 2% of the total luminescence intensity. Their temporal behavior were obtained by multiple measurements with subsequent averaging and subtraction of the interference signal comparable with the useful signal. Nevertheless, the presented temporal behavior shows a superposition

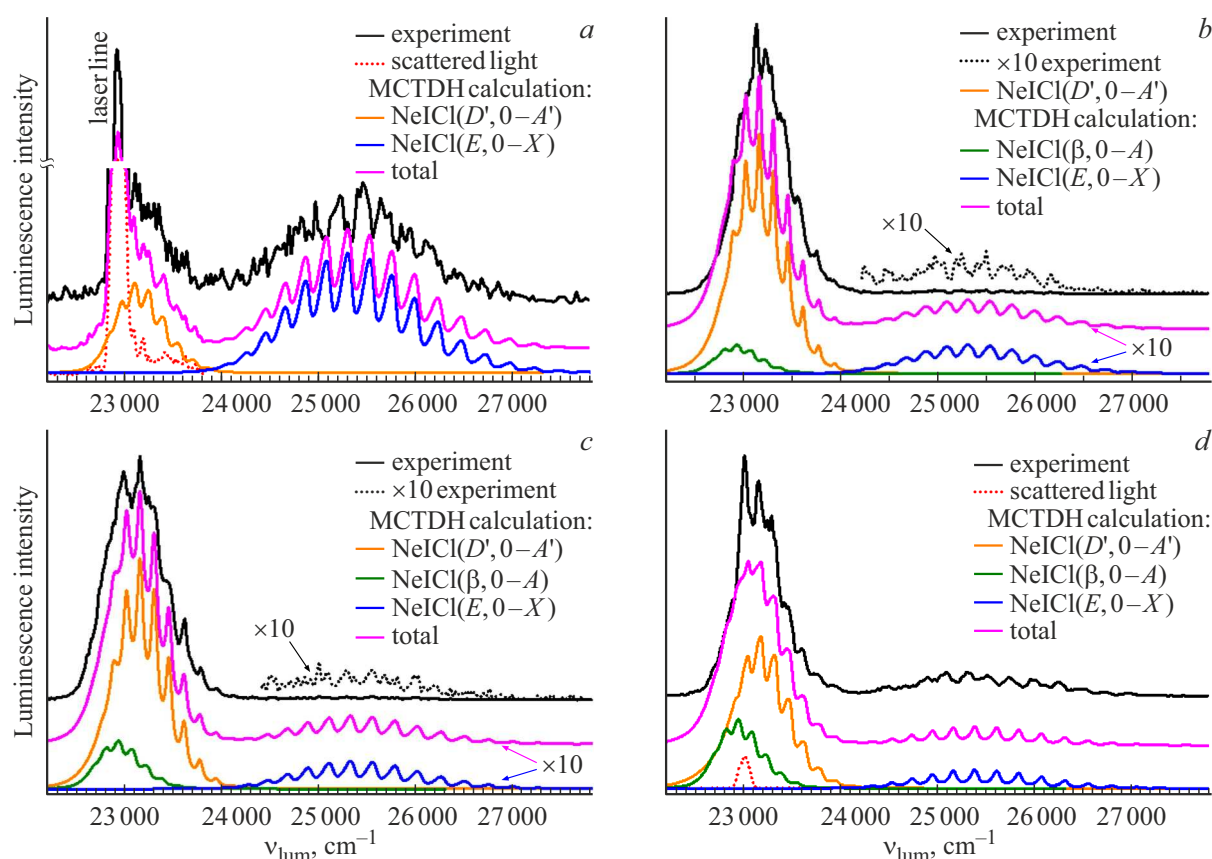


**Figure 7.** The NeICl( $E, 0, n_E \rightarrow X, D', 0 \rightarrow A'$  and  $\beta, 0, n_\beta \rightarrow A$ ) luminescence spectra measured at  $\nu_2 = 22922.8 \text{ cm}^{-1}$  (a),  $22951.8 \text{ cm}^{-1}$  (b),  $22966.3 \text{ cm}^{-1}$  (c) and  $22995.4 \text{ cm}^{-1}$  (d). Spectral resolution is FWHM = 20 Å. Simulations of the NeICl( $E, 0, n_E \rightarrow X, D', 0, n_D' \rightarrow A'$  and  $\beta, 0, n_\beta \rightarrow A$ ) spectra using shifted ICl( $E, 0 \rightarrow X, D', 0 \rightarrow A'$  and  $\beta, 0 \rightarrow A$ ) spectra of a free ICl molecule and branching ratios of the NeICl( $E, 0, n_E, D', 0, n_D'$  and  $\beta, 0, n_\beta$ ) population are given. Luminescence wavenumbers at which temporal behaviors of luminescence intensities have been measured are marked by arrows.

of two processes: (1).  $E \rightarrow X$  luminescence ( $3950 \text{ Å}$ ) of the mixed  $\beta/E/D'$  state, which similar to the time profile of luminescence at wavelengths of  $4264 \text{ Å}$  ( $D' \rightarrow A'$ ) and  $4385 \text{ Å}$  ( $\beta A$ ). (2).  $E \rightarrow X$  luminescence, the main contribution to which comes from the  $E$  state populated as a result of NeICl( $E/\beta/D'$ ) collisions with He atoms. To test this assumption, the temporal behaviors were measured at different carrier gas (He) pressures. As the pressure was reduced from 20 to 10 atm, a clear dependence on the He pressure was observed, i.e. the contribution of the luminescence due to collision-induced non-adiabatic transitions (CINATs) was significantly reduced. In the case where the contribution of the  $E$  state to the mixed NeICl( $E/\beta/D'$ ) state is large, the contribution of CINATs is small and cannot be observed (Fig. SD6, a, d). In Fig. SD6, b, c, the time dependences were approximated in the range of 80–140 ns, whereas the laser pulse ends at 50 ns. The longer luminescence decay time compared to the free molecule is due to the „pumping“ from the optically populated  $\beta/E/D'$  state. The uncertainties in determining the lifetime in this case are higher.

It is worth noting the strong  $\nu_2 = 23069.6 \text{ cm}^{-1}$  band in Fig. 4. Luminescence spectrum at this band is similar to

that of the transition in the free molecule, ICl( $E, 0 \rightarrow X$ ) (Fig. SD7). However, this band lies in the energy range within the bound part of the NeICl complex, where neither EP nor VP decay of the complex with following luminescence of free ICl( $E, 0$ ) molecules is impossible. Moreover, the radiative lifetime measured at it is 19.7(5) ns and does not correspond to the transition in a free molecule ICl( $E, 0 \rightarrow X$ ), 22.6(2) ns [13]. Therefore, we observe the luminescence of the complex itself. As noted above, luminescence spectra are determined mainly by vibrational structure of the ICl molecule. However, they are shifted because of the difference between vdW part of the PESs belonging to the  $E$  and  $X$  states. The luminescence spectrum at the  $\nu_2 = 23069.6 \text{ cm}^{-1}$  band is not shifted. It can be explained by the features of the wavefunction only. The NeICl( $E, 0, n_E$ ) vdW levels populated at this band lie close to the dissociation limit having the energy  $13 \text{ cm}^{-1}$  below it. Wavefunctions are spread over  $R, \Theta$  coordinates there. As to the NeICl( $E, 0, n_{\text{str}}, n_b \approx 0$ ) wavefunctions, they are localized on stretching vdW modes, mainly. At high vdW levels, the probability density is localized at the high  $R$  distances predominantly, where PESs of the NeICl( $E$  and  $X$ ) states are similar. In other words, this is



**Figure 8.** Experimental and calculated using the Heidelberg MCTDH code luminescence spectra of the  $\text{NeICl}(E, 0, n_E, D', 0, n_{D'}, \beta, 0, n_\beta)$  complexes for  $\nu_2 = 22922.8 \text{ cm}^{-1}$  (a),  $22951.8 \text{ cm}^{-1}$  (b),  $22966.3 \text{ cm}^{-1}$  (c) and  $22995.4 \text{ cm}^{-1}$  (d) (see Fig. 4).

a near-dissociation transition where ICl and Ne are almost separated and luminescence spectrum can be the same as the  $\text{ICl}(E \rightarrow X)$  one at our spectral resolution.

All these features suggest that: (1) the  $\text{NeICl}(E, 0, n_E, D', 0, n_{D'}, \beta, 0, n_\beta)$  luminescence spectra cannot be described by shifted spectra of a free molecule; (2) the perturbations in the  $\text{NeICl}(E, 0, n_E, D', 0, n_{D'}, \beta, 0, n_\beta)$  complexes depends strongly on their excitation energies.

### 3.5. Comparison of the experimental and calculated luminescence spectra of complexes

To calculate the luminescence spectra, we used the Heidelberg MCTDH methods [49–51] in the manner described in Ref. [15]. Experimental and calculated luminescence of the  $\text{NeICl}(E, 0, n_E, D', 0, n_{D'}, \beta, 0, n_\beta)$  complexes for  $\nu_2 = 22922.8 \text{ cm}^{-1}$  (a),  $22951.8 \text{ cm}^{-1}$  (b),  $22966.3 \text{ cm}^{-1}$  (c) and  $22995.4 \text{ cm}^{-1}$  (d) wavenumbers are given in Fig. 8, a–d.

The  $\nu_2 = 22922.8, 22941.5 \text{ cm}^{-1}$  belong to transitions to the mixed states in which the  $\text{NeICl}(E, 0, n_E = 0, 1)$  states prevail. The  $\nu_2 = 22951.8, 22966.6 \text{ cm}^{-1}$  bands belong to transitions to the mixed states in which the  $\text{NeICl}(\beta, 0, n_\beta = 0, 1)$  prevail (see Sects. IV.1, IV.4). Nev-

ertheless, the  $\text{NeICl}(D', 0 \rightarrow A')$  luminescence prevails at the  $\nu_2 = 22951.8$  and  $22966.6 \text{ cm}^{-1}$  bands; the similar feature occurs at many other excitation band which have not been analyzed in the paper. There are shifts of calculated spectra relative to the experimental ones. The similar feature has been observed in Ref. [14]. Besides, the  $\text{NeICl}(E, \nu_E = 0, n_E \rightarrow X, \nu_X, n_X)$  spectra correspond to high  $\nu_X$  values as in the  $\text{ICl}(E, \nu_E = 0 \rightarrow X, \nu_X)$  transition (see Ref. [13,15]), while the  $\text{NeICl}(X)$  PES has been calculated at  $r = r_e$ . Nevertheless, the calculated luminescence spectrum based on this PES shows reasonable agreement with the experimental data.

### 3.6. Analysis of the luminescence spectra and population and decay of the $\text{NeICl}(\text{IP})$ complexes in the energy range higher than their decay limits

All the bands presented in Fig. 5, a correspond to the energy range in which  $\text{NeICl}(E, 1, n_E, D', 1, n_{D'} \text{ and } \beta, 1, n_\beta)$  are populated. Therefore, VP and EP are possible here, and one has to observe luminescence of free  $\text{ICl}(E, 0, D', 0, \beta, 0)$  molecules. Experimental data confirm this statement: the spectra are simulated by those of free molecules, and

lifetimes are equal to those of free molecules, also (see Fig. SD8 and Ref. 13).

## 4. Conclusions

The spectroscopic characteristics of the *T*-shaped NeICl valence A1 and IP  $E0^+$ ,  $\beta 1$  states obtained in experiments and calculated using the IDIM PT1 method agree well. Experimental and calculated pump-probe and excitation spectra agree very well. We have observed the NeICl( $A$ ,  $v_A$ ,  $n_A = 0 \xleftarrow{h\nu_1} X$ ,  $0$ ,  $n_X = 1$ ) and NeICl( $E$ ,  $v_E$ ,  $\beta$ ,  $v_\beta \xleftarrow{h\nu_2} A$ ,  $v_A$ ,  $n_A = 0$ ) excitation spectra transitions only, in the experiments, and results of calculations agree very well with experimental data. The NeICl( $E$ ,  $0$ ,  $n_E \rightarrow X$ ), NeICl( $D'$ ,  $0$ ,  $n_{D'} \rightarrow A'$ ) and NeICl( $\beta$ ,  $0$ ,  $n_\beta \rightarrow A$ ) luminescence spectra can be satisfactory described by shifted spectra of a free ICl molecule. Nevertheless, these descriptions are incorrect. We have achieved satisfactory descriptions of the NeICl( $E$ ,  $0$ ,  $n_E \rightarrow X$ ), NeICl( $D'$ ,  $0$ ,  $n_{D'} \rightarrow A'$ ) and NeICl( $\beta$ ,  $0$ ,  $n_\beta \rightarrow A$ ) luminescence spectra using Heidelberg MCTDH method.

## Author Declaration

### Conflict of Interest

The authors have no conflict to discuss.

## Supplementary Material

The Supplementary material includes the NeICl( $X0^+$ ) PES reconstructed based on data of Ref. [3], contour plots of the NeICl( $A$ ,  $E$  and  $\beta$ ) PESs, pump-probe, NeICl( $A$ ,  $13$ ,  $n_A \leftarrow X$ ,  $0$ ,  $n_X$ ) (pump), ICl( $E$ ,  $0 \leftarrow A$ ,  $13$ ) (probe) and NeICl( $\beta \rightarrow A/D' \rightarrow A'$ ), NeICl( $E \rightarrow X$ ) luminescence excitation spectra, temporal behaviors of luminescence intensities measured at selected vdW modes, and luminescence spectra of the NeICl( $E$ ,  $1$ ,  $D', 1$  and  $\beta$ ,  $1$ ) complexes EP and VP products.

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